



CHALMERS

Chalmers Publication Library

COLD WATER VAPOR IN THE BARNARD 5 MOLECULAR CLOUD

This document has been downloaded from Chalmers Publication Library (CPL). It is the author's version of a work that was accepted for publication in:

Astrophysical Journal Letters (ISSN: 2041-8205)

Citation for the published paper:

Wirström, E. ; Charnley, S. ; Persson, C. (2014) "COLD WATER VAPOR IN THE BARNARD 5 MOLECULAR CLOUD". *Astrophysical Journal Letters*, vol. 788(2),

<http://dx.doi.org/10.1088/2041-8205/788/2/132>

Downloaded from: <http://publications.lib.chalmers.se/publication/200802>

Notice: Changes introduced as a result of publishing processes such as copy-editing and formatting may not be reflected in this document. For a definitive version of this work, please refer to the published source. Please note that access to the published version might require a subscription.

Chalmers Publication Library (CPL) offers the possibility of retrieving research publications produced at Chalmers University of Technology. It covers all types of publications: articles, dissertations, licentiate theses, masters theses, conference papers, reports etc. Since 2006 it is the official tool for Chalmers official publication statistics. To ensure that Chalmers research results are disseminated as widely as possible, an Open Access Policy has been adopted. The CPL service is administrated and maintained by Chalmers Library.

(article starts on next page)

COLD WATER VAPOR IN THE BARNARD 5 MOLECULAR CLOUD

E. S. WIRSTRÖM¹, S. B. CHARNLEY², C. M. PERSSON¹, J. V. BUCKLE^{3,4}, M. A. CORDINER^{2,5}, AND S. TAKAKUWA⁶

¹ Department of Earth and Space Sciences, Chalmers University of Technology, Onsala Space Observatory, SE-439 92 Onsala, Sweden; eva.wirstrom@chalmers.se

² Astrochemistry Laboratory and The Goddard Center for Astrobiology, Mailstop 691,
NASA Goddard Space Flight Center, 8800 Greenbelt Road, Greenbelt, MD 20770, USA

³ Astrophysics Group, Cavendish Laboratory, J. J. Thomson Avenue, Cambridge CB3 0HE, UK

⁴ Kavli Institute for Cosmology, Madingley Road, Cambridge CB3 0HA, UK

⁵ Institute for Astrophysics and Computational Sciences, The Catholic University of America, Washington, DC 20064, USA

⁶ Institute of Astronomy and Astrophysics, Academia Sinica, P.O. Box 23-141, Taipei 106, Taiwan

Received 2014 February 28; accepted 2014 May 21; published 2014 June 4

ABSTRACT

After more than 30 yr of investigations, the nature of gas–grain interactions at low temperatures remains an unresolved issue in astrochemistry. Water ice is the dominant ice found in cold molecular clouds; however, there is only one region where cold (~ 10 K) water vapor has been detected—L1544. This study aims to shed light on ice desorption mechanisms under cold cloud conditions by expanding the sample. The clumpy distribution of methanol in dark clouds testifies to transient desorption processes at work—likely to also disrupt water ice mantles. Therefore, the *Herschel* HIFI instrument was used to search for cold water in a small sample of prominent methanol emission peaks. We report detections of the ground-state transition of *o*-H₂O ($J = 1_{10}-1_{01}$) at 556.9360 GHz toward two positions in the cold molecular cloud, Barnard 5. The relative abundances of methanol and water gas support a desorption mechanism which disrupts the outer ice mantle layers, rather than causing complete mantle removal.

Key words: astrochemistry – ISM: individual objects (Barnard 5) – ISM: molecules – stars: formation – submillimeter: ISM

Online-only material: color figures

1. INTRODUCTION

Recent observations of interstellar H₂O by *Herschel*⁷ have revolutionized our understanding of the physics and chemistry of water at the elevated temperatures (~ 100 – 3000 K) encountered in star-forming gas (e.g., Kristensen & van Dishoeck 2011; van Dishoeck et al. 2011). In these energetic environments, the observed H₂O molecules could be formed in endoergic reactions involving H₂ or readily evaporated and/or sputtered from dust mantles.

As water is the dominant ice component in dark molecular clouds (e.g., Öberg et al. 2011), measuring its gas-phase abundance can shed light on the nature of the gas–grain interaction at low temperatures (~ 10 K)—an issue that remains unresolved after more than 30 yr of investigation (e.g., Leger et al. 1985). Water vapor has only been detected in one dark cloud core, L1544 (Caselli et al. 2012), and its low measured abundance supports the low upper limits previously obtained by SWAS (Snell et al. 2000) and ODIN (Klotz et al. 2008). This region is ostensibly the least viable target in which to search for water as many gas-phase molecules are frozen out onto the dust. Caselli et al. (2012) attribute the presence of H₂O to the photodesorption of ice molecules by a weak, ambient ultraviolet (UV) radiation field (e.g., Prasad & Tarafdar 1983).

However, this UV photodesorption mechanism cannot explain the fact that emission from putative photodesorbed molecules in dark clouds is known to be clumpy and not correlated to density structures as one would expect. Methanol in particular is a molecule that can only form by grain-surface reactions, and molecular maps show that CH₃OH is enriched

in some clumps but not in others (e.g., Buckle et al. 2006), suggesting that the ice mantles must have been removed by a transient process. Because water and methanol have similar physisorption binding energies (Sandford & Allamandola 1993), any such process should therefore also release water molecules. The gas-phase dissociative recombination of protonated water re-forms water and OH molecules (which can go on to form water again), while dissociative recombination predominantly breaks the C–O bond of protonated methanol, and so the post-desorption lifetime of H₂O molecules is much longer than that of CH₃OH ($> 10^4$ yr longer at densities of 5×10^4 cm⁻³).

In this Letter, we report on the detection of water at two positions in the cold dark cloud Barnard 5 with *Herschel* and discuss the implications for the gas–grain chemistry of molecular clouds.

2. OBSERVATIONS AND DATA REDUCTION

2.1. Barnard 5

The Barnard 5 cloud in Perseus (B5) has four known protostars, of which the most prominent Class I protostar IRS 1 drives an energetic outflow into the surrounding cloud material (Langer et al. 1996). CO and NH₃ maps indicate that B5 has a very clumpy morphology (Goldsmith et al. 1986; Pineda et al. 2011) and molecular maps in several other species (e.g., N₂H⁺, CH₃OH, HC₃N, SO, CCS, *c*-C₃H₂) show that these clumps are chemically differentiated (S. B. Charnley et al. 2014, in preparation) in a manner similar to that found in the sample of Buckle et al. (2006). We selected two positions for the *o*-H₂O search: a main methanol peak—hereafter the “methanol hotspot”—that is offset from IRS 1 by ($-2'$, $+5'$) and not associated with a core or column density peak as traced by C¹⁸O or 850 μ m continuum (Hatchell et al. 2005); and a weaker secondary peak closer to IRS 1; see Figure 1.

⁷ *Herschel* is an ESA space observatory with science instruments provided by the European-led Principal Investigator consortia and with important participation from NASA.

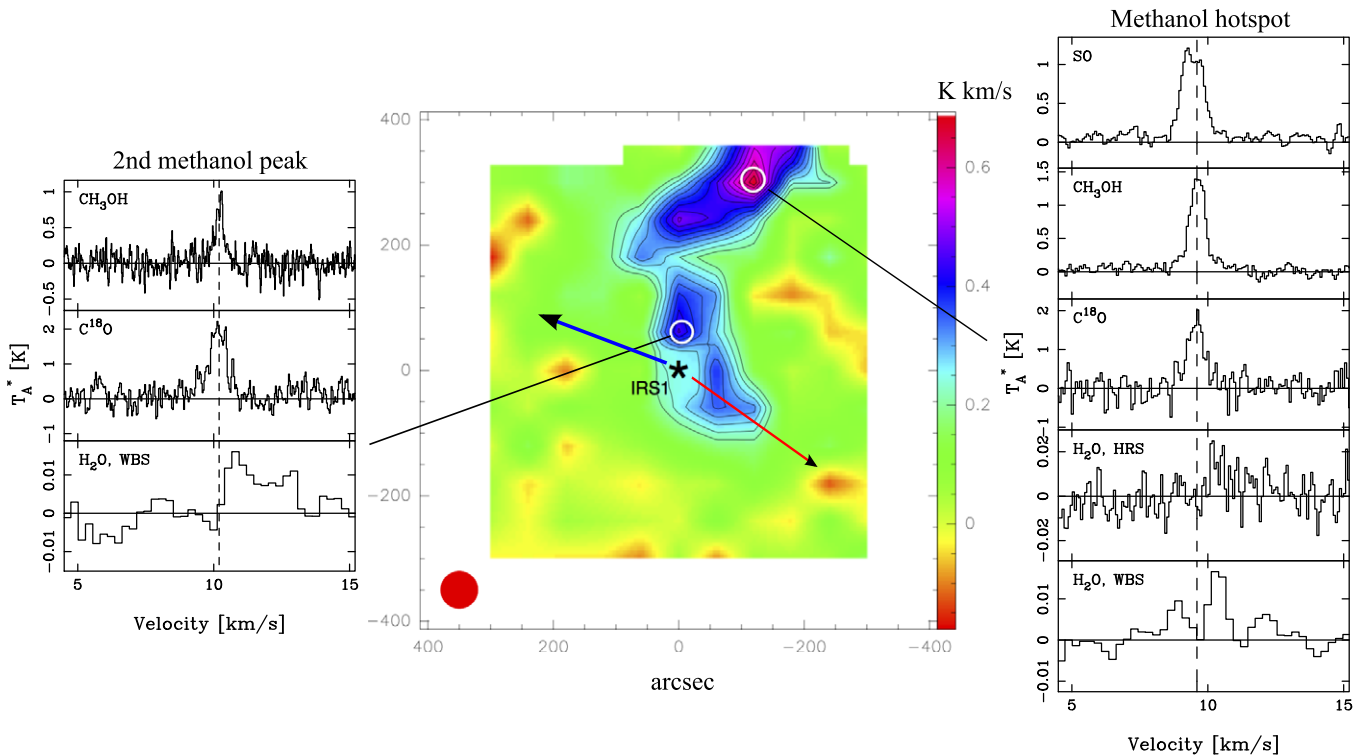


Figure 1. ARO 12 m integrated intensity map of B5 in A-CH₃OH at 96.7 GHz together with spectra toward the two observed positions, marked by white rings showing the relative size of the *Herschel* beam. Red and blue arrows show the approximate direction of the IRS 1 outflow, spatial scale is in arcseconds from IRS 1. SO and C¹⁸O were observed with the Onsala 20 m telescope (S. B. Charney et al. in preparation). Data from both spectrometers (HRS and WBS) are shown, the former redressed to a channel spacing of 0.0756 km s⁻¹ to correspond to the CH₃OH spectral resolution. The vertical dashed lines mark v_{LSR} 's as defined by the CH₃OH emission peaks (9.6 and 10.2 km s⁻¹, respectively).

(A color version of this figure is available in the online journal.)

The HIFI instrument (de Graauw et al. 2010) on board the *Herschel Space Observatory* (Pilbratt et al. 2010) was used to observe the ground-state transition of *o*-H₂O ($J = 1_{10}-1_{01}$) at 556.9360 GHz toward two positions in B5—the methanol hotspot at R.A. = 3^h47^m32^s.10, decl. = +32^d56^m43^s.0 and the second methanol peak at 3^h47^m40^s.4 + 32^d52^m28^s.0 (J2000.0)—on 2012 August 11 (OT2). The position switching mode was used with OFF positions at (−10′, 0′), chosen for lack of CO emission (Goldsmith et al. 1986). The local oscillator (LO) frequency in HIFI band 1b was set to 564.56 GHz in both L and R polarization, placing the line in the lower side band, and the total integration time (ON+OFF) was about 40 minutes for each source. Both the Wide-Band Spectrometer (WBS) and the High Resolution Spectrometer (HRS) were used, and because emission lines from the region are known to be narrow (0.5–0.8 km s⁻¹) the HRS was configured to high resolution mode with a band width of 230 MHz, corresponding to a spectral resolution of 125 kHz, or ~0.07 km s⁻¹ at this frequency. The methanol hotspot was also observed for a total of 60 minutes integration time on 2013 March 7 (Director’s Discretionary Time; DDT) using the same instrument setup. Observations were performed under observing programs OT2_ewirst01_2 and DDT_ewirst01_3, and the data presented here is available from the *Herschel* Science Archive⁸ under observing IDs 1342249424, 1342249425, and 1342266407.

The beam FWHM is 38″ at this frequency, the forward and main-beam efficiencies $\eta_1 = 96\%$ and $\eta_{\text{mb}} = 75\%$, respectively. Detailed information about the HIFI calibration including beam

efficiency, mixer sideband ratio, pointing, etc., can be found on the *Herschel* internet site.⁹ The in-flight performance is described by Roelfsema et al. (2012).

Spectra of both polarizations were reduced separately using the *Herschel* Interactive Processing Environment (Ott 2010), version 10.0.0. Subsequently, data FITS files were exported to the spectral analysis software XS¹⁰ for further reduction and analysis. After linear baseline subtraction and frequency alignment, the two polarizations for each observing ID and spectrometer were averaged together, weighted by system temperature and integration time. Pointing offsets between polarizations were within 7″, i.e., less than 20% of the beam size.

2.2. Other Sources

During the original OT2 program and the subsequent DDT_ewirst01_4, we also observed methanol peak positions in L1512, TMC-1, and TMC-1C, all offset from known protostellar activity. Table 1 summarizes these observations, with WBS rms noise temperatures on the T_{mb} scale. Only upper limits on the abundance/emission from *o*-H₂O ($J = 1_{10}-1_{01}$) were obtained. These limits and associated implications are discussed respectively in Sections 3.2 and 4.

3. RESULTS AND ANALYSIS

Two emission peaks, offset from the systemic velocity ($v_{\text{LSR}} = 9.6$ km s⁻¹) at ~8.9 km s⁻¹ and ~10.3 km s⁻¹,

⁹ <http://herchel.esac.esa.int/>

¹⁰ Developed by Per Bergman, Onsala Space Observatory, Sweden; <http://www.chalmers.se/rss/oso-en/observations/data-reduction-software>.

⁸ <http://archives.esac.esa.int/hda/ui/>

Table 1
Summary of Non-detection Observations

Source	ObsId	T_{rms}^a	$X(o\text{-H}_2\text{O})$
L1512_meth	1342250434	37 mK	$<1.5\text{e-}8$
TMC-1C, clump6-1	1342266589	9 mK	$<4\text{e-}9$
TMC-1, ch3oh-1	1342266590	8 mK	$<4\text{e-}9$

Note. ^a At 0.3 km s^{-1} resolution.

are detected toward the methanol hotspot in the combined WBS spectrum from both observing runs. The rms noise at this resolution is 2.9 mK and the confidence level of the detection across a 3 km s^{-1} range around the v_{LSR} is 6σ . Comparing the water spectra to those of other molecules toward the methanol hotspot in Figure 1, the water line profile can be interpreted as a self-absorbed emission line around the systemic velocity. The asymmetry in peak intensities, with a suppressed blue component, is then indicative of low velocity expansion or outflow in the methanol clump. This is commonly observed for water in massive star-forming regions (e.g., Ashby et al. 2000), but has never before been observed toward cold, non-star-forming gas.

Toward the second methanol peak, the WBS spectrum noise rms is 5 mK. We find a tentative detection of water emission at velocities higher than the v_{LSR} , corresponding to 3σ rms when integrated over a 3 km s^{-1} range around the v_{LSR} ; see Figure 1. See below for a further discussion on this tentative detection.

Before further analysis, water spectra were multiplied by the total efficiency factor of $\eta_i/\eta_{\text{mb}} = 1.26$ (HIFI Observers' Manual version 2.4) in order to get them on the T_{mb} scale. Note that the HRS OT2 data show artifacts in the spectrum at the velocity of the emission feature and is therefore not included in any analysis.

3.1. Analysis

In recent years, submillimeter continuum surveys of the Perseus region have contributed to the understanding of cold dust properties around the B5 methanol hotspot. A dust temperature map based on *Spitzer* and Two Micron All Sky Survey data is available from the COMPLETE database¹¹ and estimates $T_{\text{dust}} = 13$ and 14 K for the methanol hotspot and the second position in B5, respectively (Schnee et al. 2008). In addition, the whole Perseus region has been observed with both PACS and SPIRE (S. Pezzuto et al. in preparation) as part of the Gould Belt Survey *Herschel* Key program (KPGT_pandre_1, André et al. 2010). Temperature and column density maps show no column density peak coinciding with the methanol hotspot within the *Herschel* main beam, while the dust temperature agrees with the Schnee et al. (2008) map (S. Pezzuto 2012, private communication). The second methanol peak however coincides to within $20''$ with a column density peak and temperature minimum (12 K), indicating the presence of a core.

The $1_{10}\text{-}1_{01}$ emission line becomes optically thick at low column densities, and in cold gas essentially all *ortho*-water molecules reside in their ground state. However, water excitation and radiative transfer are significantly complicated by the very large optical depths encountered, subthermal excitation, and possibility of population inversion. In order to interpret the water line profile suggested by our observations we use an accelerated lambda iteration (ALI) scheme (Rybicki & Hummer 1991), including these non-LTE effects, to solve the radiative

Table 2
ALI Cloud Model Properties

	CH ₃ OH "hotspot"	<i>o</i> -H ₂ O "hotspot"	<i>o</i> -H ₂ O 2nd B5 pos
Clump			
R_{max} (pc)	0.04	0.04	0.04
$n(\text{H}_2)$ (cm^{-3})	5.0(4)	5.0(4)	9.0(4)
M (M_{\odot})	0.64	0.64	1.35
T_{kin} (K)	9	9	10
T_{dust} (K)	13	13	14
v_{turb} (km s^{-1})	0.3	0.42	0.5
$v_{\text{exp,max}}$ (km s^{-1})	...	0.12	0.5
X_{mol}	3.9(−8)	1.9(−8)	5.5(−9)
Envelope			
R_{max} (pc)	...	0.08	0.08
$n(\text{H}_2)$ (cm^{-3})	...	5.0(3)	9.0(3)
M (M_{\odot})	...	0.55	0.79
v_{exp} (km s^{-1})	...	0.12	0.5

Note. 5.0(4) means 5.0×10^4 , etc.

transfer in a spherically symmetric model cloud. The ALI code is based on that used and tested by, e.g., Maercker et al. (2008), but modified by P. Bergman to take into account several collision partners.

To estimate the physical properties of the desorbed gas at the B5 methanol hotspot, the observed methanol emission line quartet around 96.7 GHz (S. B. Charnley et al. in preparation) was modeled using ALI. A- and E-type methanol were modeled separately, with molecular data from the Cologne Database for Molecular Spectroscopy¹² (Müller et al. 2005) and collision rates adapted from Rabli & Flower (2010). Both line shapes and intensities are well-reproduced by a homogeneous, 0.08 pc diameter spherical model clump whose physical properties are given in Table 2, and where the molecular spin ratio A/E ~ 1 . The size of the methanol emitting region in this model is defined by map data, the total H₂ column density is of the same order as that found by the Gould Belt Survey, but the gas kinetic temperature has to be lower than their derived dust temperature, as low as 9 K, to reproduce observed line intensity ratios. The resulting average column density over the 64'' Arizona Radio Observatory (ARO) 12 m beam is $N(\text{CH}_3\text{OH}) = 1.3 \times 10^{14} \text{ cm}^{-2}$, and the total H₂ mass of the model clump is $0.64 M_{\odot}$, which is less than half of its virial mass ($3.02 M_{\odot}$) indicating that it is gravitationally unbound.

Molecular data for *o*-H₂O was taken from the Leiden Atomic and Molecular Database¹³ (Schöier et al. 2005). Collisional excitation of *o*-H₂O by both *ortho*- and *para*-H₂ is taken into account, state-to-state collisional rates are adapted from Dubernet et al. (2009) and Daniel et al. (2011), respectively. At low temperatures water excitation is very sensitive to the *ortho*-to-*para* ratio (OPR) in H₂ and the thermal equilibrium OPR at 10 K is 3.6×10^{-7} . However, due to formation on grains and chemical recycling of H₂, such low OPR is not expected to be reached within the lifetime of a dark cloud (e.g., Flower et al. 2006; Pagani et al. 2011), which is also consistent with observations (Troscompt et al. 2009). Here we adopt a constant OPR value of 0.001, close to that predicted for a 10 K cloud of H₂ density 10^4 cm^{-3} by Flower et al. (2006).

Peak intensities in the water line profile toward the B5 methanol hotspot can be modeled in ALI from the same

¹¹ <http://www.cfa.harvard.edu/COMPLETE/>

¹² <http://www.ph1.uni-koeln.de/vorhersagen/>

¹³ Available at <http://www.strw.leidenuniv.nl/~moldata>.

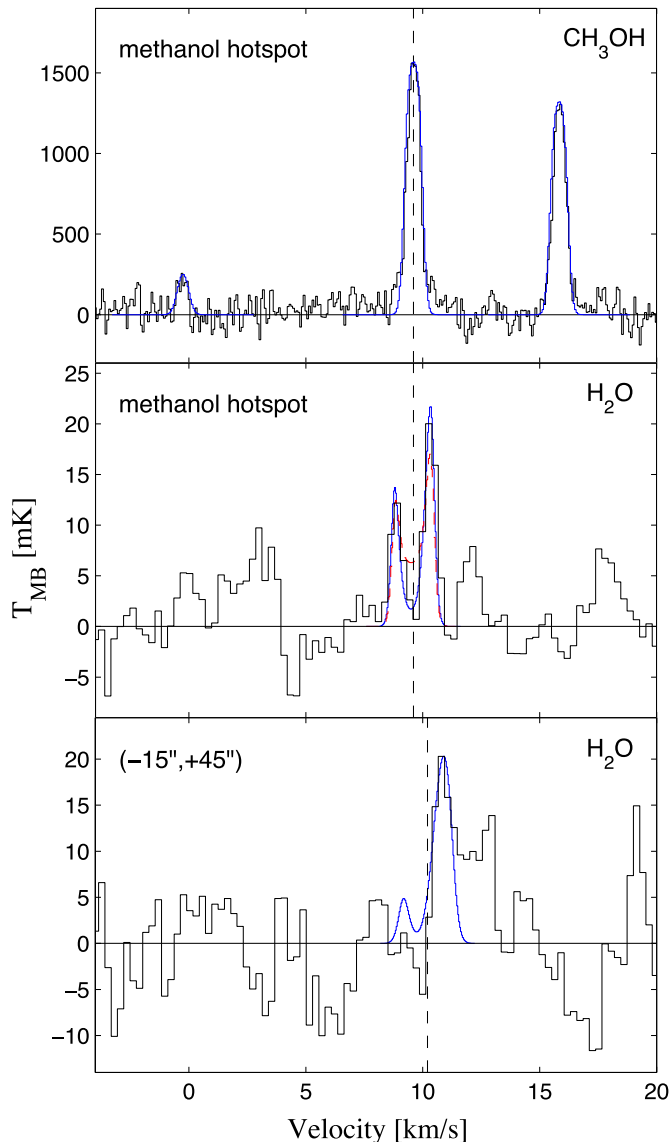


Figure 2. ALI model results of slowly expanding spherical clouds (in blue; same parameters as in Table 2) as compared to observed spectral features (black) toward the two positions in B5. Toward the methanol hotspot, the red dashed curve shows modeled water emission when excluding the envelope component. Local v_{LSR} 's are marked by dashed vertical lines.

(A color version of this figure is available in the online journal.)

spherical clump model as used for methanol above by adjusting the turbulent velocity and adding a slow expansion increasing linearly from zero at the cloud center to $v_{\text{exp,max}}$ at the cloud edge. This slow expansion velocity can be included in the methanol model without hampering the fit, see the top panel of Figure 2. However, in order to reproduce the deep central absorption, an outer envelope of lower density has to be included in the water model. The middle panel of Figure 2 compares the best-fit model with envelope to the corresponding model without, where the temperature, turbulent velocity, and water abundance are kept constant, and the expansion velocity levels out to a constant value in the envelope (parameters in the left column of Table 2). Note that in the best-fit model the bulk of the water column originates in the methanol clump while the methanol-free envelope, possibly testifying to a longer time since desorption, only contributes by $\sim 10\%$. At a temperature of 9 K we find that the ortho water abundance, $X_{o\text{-H}_2\text{O}}$, has to

be around 2×10^{-8} . At a 10 times higher H_2 OPR, this water abundance would only have to be $\sim 5\%$ lower for the model to fit the data.

Through only small adjustments to the cloud parameters, the water spectrum toward the second methanol peak may also be reproduced using the ALI model. The lower panel of Figure 2 shows this WBS spectrum together with the model result for a somewhat denser ($n(\text{H}_2) = 10^5 \text{ cm}^{-3}$) cloud at $v_{\text{LSR}} = 10.2 \text{ km s}^{-1}$ where the expansion velocity increases to be equal to the turbulent velocity in the outer envelope (see parameters in Table 2). This velocity relation is required to explain the lack of a blue-shifted emission component as seen toward the methanol hotspot. The temperature is kept at 10 K and the ortho water abundance in this model is 5.5×10^{-9} relative to H_2 .

3.2. Other Sources

For a first-order estimate of the L1512, TMC-1, and TMC-1C water abundance upper limits we exploit the fact that emission in the $o\text{-H}_2\text{O}$ ground-state transition never gets thermalized at typical dark cloud densities; therefore, even if the emission is optically thick the high escape probability for emitted photons ensures that we are in the optically thin limit (e.g., Snell et al. 2000). Then the $o\text{-H}_2\text{O}$ abundance can be expressed as

$$X(o\text{-H}_2\text{O}) = 6.0 \times 10^8 \frac{\int T_{\text{R}} dv}{C(T_{\text{K}}) n_{\text{H}_2} N(\text{H}_2)} \exp(26.8/T_{\text{K}}), \quad (1)$$

where the integrated intensity is given in K km s^{-1} , the column density in cm^{-2} , and the density in cm^{-3} . Table 1 lists the resulting water abundance upper limits, assuming a kinetic temperature of $T_{\text{K}} = 10 \text{ K}$, a molecular hydrogen density of $5 \times 10^4 \text{ cm}^{-3}$, and H_2 column density of 10^{22} cm^{-2} .

4. DISCUSSION

We have detected cold gas-phase water at two positions in the B5 molecular cloud. The fact that these positions correspond to peaks in the CH_3OH distribution supports the view that the water has recently been desorbed from dust grain mantles. The derived $\text{H}_2\text{O}/\text{H}_2$ abundance of $\approx 2.5 \times 10^{-8}$ at the methanol ‘‘hotspot’’ is more than an order of magnitude greater than that found by Caselli et al. (2012) in L1544, whereas that derived closer to IRS 1 of $\approx 5 \times 10^{-9}$ is comparable. The large amount of water and methanol observed at a position uncorrelated to any core-like density structure suggests that UV photodesorption is not the mechanism for removing molecules from the surfaces of dust grains, but supports the scenario modeled by Charney & Rodgers (2009), where spatial inhomogeneity is a consequence of the ice formation process.

The solid CH_3OH fraction toward B5 IRS 1 has been estimated to be $< 6\%$ relative to water ice and so, if evaporated, the water in B5 would have $\text{H}_2\text{O}/\text{H}_2 \approx 10^{-4}$ (Boogert et al. 2004). Even considering a several orders of magnitude lower water ice abundance further away from the cloud density peak, the low observed abundance of gas-phase water at the methanol hotspot implies that the operating desorption mechanism cannot cause complete disruption of ice mantles. For a derived $\text{CH}_3\text{OH}/\text{H}_2$ abundance of $\approx 4 \times 10^{-8}$ at the methanol hotspot, the gas-phase $\text{CH}_3\text{OH}/\text{H}_2\text{O}$ ratio in the clump is 1.5. This demonstrates that only partial removal of ice material rich in methanol has occurred. This is consistent with recent theoretical models of grain mantle formation which follow the chemical composition

of each ice monolayer as it is formed (Charnley & Rodgers 2009; Garrod & Pauly 2011; Vasyunin & Herbst 2013). In these calculations CH₃OH-rich monolayers form late in the evolution, as the atomic O/H ratio in the gas falls and water formation becomes concomitantly less efficient, and resides nearest to the mantle surface. Desorption of these monolayers can account for the CH₃OH/H₂O ratios measured in B5 and therefore places constraints on the efficiency of the actual mantle desorption process.

Ice mantle desorption at the methanol hotspot is not likely related to the protostellar activity of IRS 1: the large-scale outflow extends in the NE–SW directions rather than to the NW (Yu et al. 1999), and the projected distance from IRS 1 of ~ 0.55 pc (see Figure 1) is too far for magnetohydrodynamic (MHD) waves to propagate and cause mantle disruption (e.g., Markwick et al. 2000). If cosmic-ray-induced photoevaporation or exoergic surface reactions were dominating the desorption, the methanol distribution would be less clumpy and would follow the H₂ column density distribution. Thus such continuous processes cannot explain the elevated levels of desorbed ices at the hotspot.

One possibility is that desorption is caused by collisions between small gas clumps that are interacting and merging (Takakuwa et al. 2003; Buckle et al. 2006). In this scenario, collisions between individual grains cause a transient heating which can result in runaway recombination of free radicals in the ice and subsequent sublimation of ice mantle molecules (Schutte & Greenberg 1991). For collision velocities of the same order as the methanol line width of ~ 0.7 km s⁻¹, colliding grain temperatures would rise to $\gtrsim 90$ K (following Draine 1985), warm enough to thermally desorb water and methanol. Evaporation of a small fraction of the outer ice layers in the form of H₂O or CH₃OH would rapidly cool these grains down to < 25 K and thus be consistent with the observed low dust temperatures and gas-phase abundances of water and methanol as compared to the ice toward IRS 1.

As mentioned, the second methanol peak nearly coincides with a column density peak and temperature minimum (12 K) in continuum, indicating the presence of a core, and it is only about 0.07 pc away from IRS 1. It also exhibits a lower methanol and water abundance than the methanol hotspot—water abundance is close to that observed toward L1544 (Caselli et al. 2012). Therefore UV photodesorption of water in a layer around the core; similar to the L1544 model; cannot be excluded, but the position is also close enough to IRS 1 for propagating MHD waves to cause mantle disruption.

Can the fact that water is not detected in this scenario? In the two TMC 1 sources the estimated gas-phase CH₃OH/H₂O ratios, based on methanol data from Takakuwa et al. (2000, 2003), are at least five times higher than in B5. Thus, the desorbed outer monolayers would have to reflect that ratio. However, if water line self-absorption is as severe as in B5, the upper limits given in Table 1 could be adjusted upward by as much as a factor of five, accounting for that difference. The upper limit in L1512 is not significant assuming that similar mantle removal processes are at work.

The discovery of gas-phase water in B5 brings new insight to the enigma of interstellar water chemistry. In B5, future observations of other complex mantle molecules believed to form along with methanol, as well as various deuterated isotopologues, will shed further light on the nature of the surface desorption process.

Part of this work was supported by NASA’s Exobiology Program and The Goddard Center for Astrobiology. E.S.W. and C.M.P. acknowledge generous support from the Swedish National Space Board.

Facility: Herschel

REFERENCES

- André, P., Men’shchikov, A., Bontemps, S., et al. 2010, *A&A*, **518**, L102
 Ashby, M. L. N., Bergin, E. A., Plume, R., et al. 2000, *ApJL*, **539**, L115
 Boogert, A. C. A., Pontoppidan, K. M., Lahuis, F., et al. 2004, *ApJS*, **154**, 359
 Buckle, J. V., Rodgers, S. D., Wirstrom, E. S., et al. 2006, *FaDi*, **133**, 63
 Caselli, P., Keto, E., Bergin, E. A., et al. 2012, *ApJL*, **759**, L37
 Charnley, S. B., & Rodgers, S. B. 2009, in ASP Conf. Ser. 420, *Bioastronomy 2007: Molecules, Microbes and Extraterrestrial Life*, ed. K. J. Meech, J. V. Keane, M. J. Mumma, J. L. Siefert, & D. J. Werthimer (San Francisco, CA: ASP), **29**
 Daniel, F., Dubernet, M.-L., & Grosjean, A. 2011, *A&A*, **536**, A76
 de Graauw, T., Helmich, F. P., Phillips, T. G., et al. 2010, *A&A*, **518**, L6
 Draine, B. T. 1985, in *Protostars and Planets II*, ed. D. C. Black & M. S. Matthews (Tucson, AZ: Univ. Arizona Press), **621**
 Dubernet, M.-L., Daniel, F., Grosjean, A., & Lin, C. Y. 2009, *A&A*, **497**, 911
 Flower, D. R., Pineau Des Forets, G., & Walmsley, C. M. 2006, *A&A*, **449**, 621
 Garrod, R. T., & Pauly, T. 2011, *ApJ*, **735**, 15
 Goldsmith, P. F., Langer, W. D., & Wilson, R. W. 1986, *ApJL*, **303**, L11
 Hatchell, J., Richer, J. S., Fuller, G. A., et al. 2005, *A&A*, **440**, 151
 Klotz, A., Harju, J., Ristorcelli, I., et al. 2008, *A&A*, **488**, 559
 Kristensen, L. E., & van Dishoeck, E. F. 2011, *AN*, **332**, 475
 Langer, W. D., Velusamy, T., & Xie, T. 1996, *ApJL*, **468**, L41
 Leger, A., Jura, M., & Omont, A. 1985, *A&A*, **144**, 147
 Maercker, M., Schöier, F. L., Olofsson, H., Bergman, P., & Ramstedt, S. 2008, *A&A*, **479**, 779
 Markwick, A. J., Millar, T. J., & Charnley, S. B. 2000, *ApJ*, **535**, 256
 Müller, H. S. P., Schlöder, F., Stutzki, J., & Winnewisser, G. 2005, *JMoSt*, **742**, 215
 Öberg, K. I., Boogert, A. C. A., Pontoppidan, K. M., et al. 2011, *ApJ*, **740**, 109
 Ott, S. 2010, in ASP Conf. Ser. 434, *Astronomical Data Analysis Software and Systems XIX*, ed. Y. Mizumoto, K.-I. Morita, & M. Ohishi (San Francisco, CA: ASP), **139**
 Pagani, L., Roueff, E., & Lesaffre, P. 2011, *ApJL*, **739**, L35
 Pilbratt, G. L., Riedinger, J. R., Passvogel, T., et al. 2010, *A&A*, **518**, L1
 Pineda, J. E., Goodman, A. A., Arce, H. G., et al. 2011, *ApJL*, **739**, L2
 Prasad, S. S., & Tarafdar, S. P. 1983, *ApJ*, **267**, 603
 Rabli, D., & Flower, D. R. 2010, *MNRAS*, **406**, 95
 Roelfsema, P. R., Helmich, F. P., Teyssier, D., et al. 2012, *A&A*, **537**, A17
 Rybicki, G. B., & Hummer, D. G. 1991, *A&A*, **245**, 171
 Sandford, S. A., & Allamandola, L. J. 1993, *ApJ*, **417**, 815
 Schnee, S., Li, J., Goodman, A. A., & Sargent, A. I. 2008, *ApJ*, **684**, 1228
 Schöier, F. L., van der Tak, F. F. S., van Dishoeck, E. F., & Black, J. H. 2005, *A&A*, **432**, 369
 Schutte, W. A., & Greenberg, J. M. 1991, *A&A*, **244**, 190
 Snell, R. L., Howe, J. E., Ashby, M. L. N., et al. 2000, *ApJL*, **539**, L93
 Takakuwa, S., Kamazaki, T., Saito, M., & Hirano, N. 2003, *ApJ*, **584**, 818
 Takakuwa, S., Mikami, H., Saito, M., & Hirano, N. 2000, *ApJ*, **542**, 367
 Troscompt, N., Faure, A., Maret, S., et al. 2009, *A&A*, **506**, 1243
 van Dishoeck, E. F., Kristensen, L. E., Benz, A. O., et al. 2011, *PASP*, **123**, 138
 Vasyunin, A. I., & Herbst, E. 2013, *ApJ*, **762**, 86
 Yu, K. C., Billawala, Y., & Bally, J. 1999, *AJ*, **118**, 2940